

gram to mole:

(Forward)

(Given)

g | 1 mole

g

(molar mass)
periodic
table

mole to gram:

(Backward)

(Given)

mole

(molar mass)

g

1 mole

Gram to Atoms

(Forward)

(Given in problem)

g	1 mol	6.02×10^{23} atoms
	g	1 mol

(molar mass: use periodic table)

(Given)

atoms	1 mol	(molar mass) g
	atoms	1 mol

6.02×10^{23}

(Backward)