

Types of reactions (and how to tell them apart)

Double Displacement;
⊕ and ⊖ switch places

Examples of:

Single Displacement;
There is a "loner" atom; ⊕ and ⊖ switch places

Synthesis (Combination)
Two reactants combine to make (1) product

Decomposition
One reactant makes 2 or more products (S)

Combustion